

## V. Klasse. Carbonate und Nitrate

Abteilung N: Nitrate

| Mineral      | Formel   | Zusammensetzung in Masse-% |       |                        |       |                        |       |                        |       |
|--------------|--|----------------------------|-------|------------------------|-------|------------------------|-------|------------------------|-------|
|              |  |                            |       |                        |       |                        |       |                        |       |
| Nitronatrium | $\text{NaNO}_3$  | $\text{Na}_2\text{O}$      | 36,46 | $\text{N}_2\text{O}_5$ | 63,54 |                        |       |                        |       |
| Nitrokalium  | $\text{KNO}_3$   | $\text{K}_2\text{O}$       | 46,59 | $\text{N}_2\text{O}_5$ | 53,41 |                        |       |                        |       |
| Nitrobarium  | $\text{Ba}(\text{NO}_3)_2$                               | $\text{BaO}$               | 58,67 | $\text{N}_2\text{O}_5$ | 41,33 |                        |       |                        |       |
| Nitrocalcium | $\text{Ca}(\text{NO}_3)_2 \cdot 4 \text{H}_2\text{O}$    | $\text{CaO}$               | 23,75 | $\text{H}_2\text{O}$   | 30,51 | $\text{N}_2\text{O}_5$ | 45,74 |                        |       |
| Gerhardtium  | $\text{Cu}_2((\text{OH})_3/\text{NO}_3)$                 | $\text{CuO}$               | 66,25 | $\text{H}_2\text{O}$   | 11,25 | $\text{N}_2\text{O}_5$ | 22,49 |                        |       |
| Likasit      | $\text{Cu}_6((\text{OH})_7(\text{NO}_3)_2(\text{PO}_4))$ | $\text{CuO}$               | 66,35 | $\text{H}_2\text{O}$   | 8,77  | $\text{P}_2\text{O}_5$ | 9,87  | $\text{N}_2\text{O}_5$ | 15,02 |